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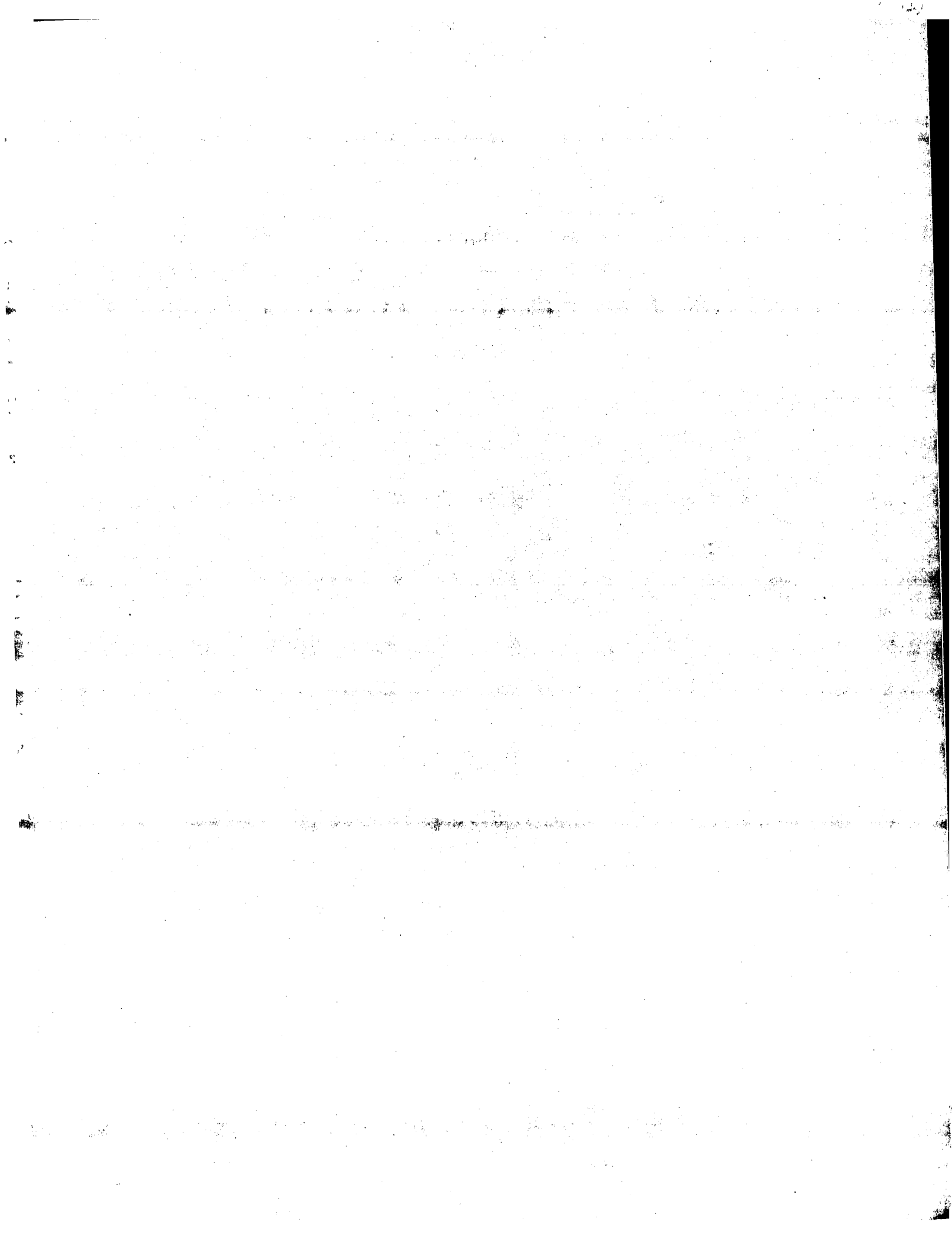
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**EP 0 295 903 B1**

## Description

The present invention relates to hair preparations having a high sebum absorptivity and capable of making the users comfortable during the use and of reducing greasiness of the scalp and hair.

5 Then it relates to hair preparations comprising an anti-dandruff agent and a specified oil-absorptive substance being capable of absorbing sebum.

Although sebum makes the skin and hair glossy and smooth when the secretion thereof is proper, it makes the skin and hair greasy and sticky when secreted excessively. Further, a makeup applied to the face is impaired by the sebum. In addition, the apparent volume of the hair is reduced by the sebum to  
10 make the keeping of a beautiful hairstyle impossible. Although the sebum secreted excessively can be easily removed by washing, the amount thereof becomes excessive again within several hours on the skin and within several days on the scalp and the hair and, therefore, repeated washing is necessitated.

Under these circumstances, studies are in progress for the purpose of physiologically controlling the secretion of the sebum or physically absorbing the excessive sebum with a powder.

15 However, the former method is not employed practically, since no safe and effective medicine has been found yet.

Although the latter method wherein the sebum is physically absorbed by a powder has no serious problem of safety, it has still problems that the sebum absorptivity is insufficient and that the users do not feel comfortable during the use.

20 For example, a process for controlling the secretion of sebum with an inorganic powder such as a clay mineral or an oxide is described in Japanese Patent Laid-Open No. 65807/1986. However, porous or bedded inorganic substances usually have a defect that the oil absorptivity thereof is seriously reduced by sweat or dirt.

A method of reducing seborrhea with silylated silica gel is described in Japanese Patent Laid-Open No.  
25 144710/1984. In this method, the surfaces of inorganic powdery particles are made hydrophobic and the sebum is absorbed in interparticle spaces of aggregates of the primary particles. No high oil absorptivity can be expected in this method.

U.S. Patent No. 4,489,058 discloses a method of absorbing sebum by a copolymer of styrene with stearyl methacrylate having a high compatibility with sebum by swelling in order to keep the skin from acne.  
30 Although this method is excellent in that the sebum can be absorbed even in the presence of sweat or dirt, the absorption is yet insufficient and the touch after absorption of the sebum is poor, since the sebum is absorbed by mere swelling. Another defect of this method is that the copolymer difficulty adheres to hair, since it has an average particle diameter of around 150  $\mu\text{m}$ .

Various anti-dandruff agents are incorporated in hair preparations such as shampoos, rinses and hair  
35 tonics in order to prevent dandruff formation or itch. All of them have an antibiotic activity and activities of dissolving a corneous layer or controlling the metabolic rate of this layer. As a result, the dandruff formation on the scalp is prevented. However, it has been pointed out that daily use of the anti-dandruff agent exerts a significant effect on the physical properties of hair. Namely, hair becomes greasy and sticky soon after shampooing [see, for example, International Journal of Cosmetic Science, 5, 77 (1983)].

40 The mechanism of the phenomenon where hair becomes greasy after the daily use of the antidandruff agent-containing composition has not been fully elucidated yet. In this connection, no anti-dandruff composition having also an effect of inhibiting or retarding the realization of greasiness has been developed.

As for special oil-absorptive substances capable of absorbing sebum, for example, colloidal silica is described in Japanese Patent Publication No. 20793/1971 and a combination of a hydroxycholanic acid  
45 derivative and a powdery oil absorbent is disclosed in European Patent No. 58000. They are used as raw material for the skin cosmetics. However, no anti-dandruff composition usable for controlling or retarding the realization of greasiness has been reported.

After clinical analysis of the behavior of sebum to diffuse toward the hair as a result of the repeated use of ordinary anti-dandruff agents made for the purpose of overcoming the defect that the hair becomes  
50 greasy at an early stage when the ordinary anti-dandruff agent is used every day, the inventors have found that the secreted sebum is distributed to the surface of the scalp and then to the hair after the laps of days, more precisely the sebum reaches saturation on the surface of the scalp quickly in one or two days and then the diffusion thereof toward the hair begins, that the fragments of a corneous layer on the surface of the scalp, i.e., dandruff, play an important role as a sebum reservoir structure, that a decrease in the  
55 dandruff due to the effect of the anti-dandruff agent brings about a decrease in the amount of the sebum stored on the scalp surface and an increase, on the contrary, in the diffusion rate of the sebum toward hair, and that this phenomenon is remarkable one day after shampooing and thereafter.

After intensive investigations made for the purpose of overcoming the above-described defects, the inventors have found that the problems can be solved by using a particular porous, oil-absorptive vinyl polymer. The present invention has been completed on the basis of this finding.

The inventors have supposed that when a new sebum reservoir structure is formed as a substitute for the fragment of a corneous layer, which are unfavorable from the viewpoint of beauty, on the scalp surface and/or hair surface, the development of the greasiness can be delayed or eliminated. After intensive investigations made for the purpose of developing a means of forming the sebum reservoir structure, the inventors have found that when an anti-dandruff composition comprising a carrier far smaller than said fragment and having a sebum absorptivity higher than that of said fragment is used, the dandruff and itch can be prevented, controlled and/or relieved and, in addition, the development of the greasiness can be retarded, while this problem could not be solved in the prior art.

The present invention has been completed on the basis of the above-described findings. The present invention provides new hair preparations derived from a new idea unlike ordinary anti-dandruff compositions. The new hair preparations have a remarkable effect of preventing or removing dandruff and itch and controlling the greasiness of hair.

The invention provides a hair preparation characterised in that it comprises particles of an oil-absorptive polymer selected from polymethacrylic stearate, polyacrylic stearate, a methacrylic stearate/styrene copolymer and an acrylic stearate/styrene copolymer having a solubility parameter of 7 to 10, said particles having an average particle diameter of 0.0005 to 30  $\mu\text{m}$  (microns) and a pore volume of at least 0.1 ml/g, said polymer having a squalene absorption of at least 1 ml/g, and a carrier.

1 micron is equal to 1 micrometer ( $\mu\text{m}$ ).

It is preferable that the hair preparation comprises 0.05 to 5 percent by weight of the oil-absorptive polymer and the balance of the carrier. It is also preferred that the oil-absorptive polymer having an average particle size in the range 0.005 to 30 microns has such a size distribution that it contains 90 percent by weight or more of those having particle sizes in that range.

The oil-absorptive polymer defined above is prepared preferably by dissolving the monomer in a nonpolymerizable organic solvent, suspending, dispersing or emulsifying the solution in water, conducting the polymerization and thereafter removing the organic solvent.

The polymerization can be conducted by any known method such as suspension, dispersion or emulsion polymerization. Suspension polymerization is recommended when the polymer having a particle diameter of 10  $\mu\text{m}$  or larger is desired. Dispersion polymerization and emulsion polymerization are recommended when the polymer having particle diameters of 1 to 10  $\mu\text{m}$  and 1  $\mu\text{m}$  or less, respectively, is desired. From the viewpoints of the touch of the skin and hair after application of the preparation and adhesion of the particles, smaller particles are preferred. However, emulsion polymerization has a defect that a surfactant used in a relatively large amount therein cannot be removed easily.

The polymer can be made porous by removing nonpolymerizable substances such as an organic solvent, plasticizer and linear polymer contained in the polymerisation solution after the completion of the polymerization. A particularly preferred process comprises dissolving the monomer in a nonpolymerizable organic solvent which is a good solvent for the monomer but a poor solvent for the polymer formed, suspending, dispersing or emulsifying the monomer solution in water, polymerizing it and removing the organic solvent after the completion of the polymerization.

If the solubility parameter of the vinyl monomer is outside the range of 7 to 10, the compatibility of the product with sebum is poor, and therefore, no sufficient sebum absorptivity can be obtained.

These monomers can be used either alone or in the form of a combination of two or more of them.

When stearyl acrylate or methacrylate is used singly, the polymer will have a low glass transition point  $T_g$  and a low porosity, though it has a high compatibility with sebum. In such a case, the monomer is preferably copolymerized with a comonomer capable of forming a polymer having a  $T_g$  of 100°C (373°K) or higher, such as styrene.

Particularly preferred monomers are stearate esters, with acrylic or methacrylic acid. Although they can be homopolymerized, it is preferred to copolymerize the monomer with styrene in order to further increase the porosity and oil absorptivity (refer to Preparation Examples 1 and 2). The weight ratio of the acrylate or methacrylate to styrene to be copolymerized is preferably at least 30/70. When this ratio is less than 30/70, the polymer becomes hard and, therefore, the touch thereof becomes poor unfavorably.

To obtain an excellent touch, it is preferred that the oil-absorptive polymer according to the present invention is swellable with sebum but insoluble therein. Therefore, the polymer is preferably crosslinked. Crosslinking is conducted by a process wherein a polyfunctional monomer is added during the polymerisation, a crosslinking process or a self-crosslinking process. The polyfunctional monomers include, for example, polyvinyl aromatic compounds such as divinylbenzene, polyvinyl heterocyclic compounds such as

divinylpyridine, dimethacrylates such as ethylene glycol dimethacrylate and triethylene glycol dimethacrylate, and diacrylates such as ethylene glycol diacrylate and triethylene glycol diacrylate. The amount of the crosslinking agent is preferably 50 wt.% or below based on the polymer. When it exceeds 50 wt.%, the swellability of the polymer is seriously inhibited. A particularly preferred amount of the crosslinking agent is 0.01 to 50 wt.% based on the total polymer.

The organic solvents usable for making the polymer porous in the present invention include, for example, aromatic compounds such as toluene and benzene, esters such as ethyl acetate and butyl acetate, alcohols such as isoamyl alcohol and methylisobutyl carbinol, saturated hydrocarbons such as n-hexane, n-octane and n-dodecane, and halogenated solvents such as dichloroethane and trichloroethylene. Among them, those which are good solvents for the vinyl monomers having a solubility parameter of 7 to 10 and which are poor solvents for the polymers prepared from the monomers in the present invention are particularly preferred. The preferred solvents include, for example, aliphatic hydrocarbons such as hexane, octane and dodecane, and aromatic hydrocarbons such as toluene and benzene. The weight ratio of the monomer to the organic solvent is preferably 1/1 to 1/4.

The plasticizers are, for example, dioctyl phthalate and dibutyl adipate. The linear polymers are, for example, polystyrene and polyvinyl acetate.

The polymerization initiators are ordinarily used oil-soluble ones. They include, for example, peroxides and azo compounds such as benzoyl peroxide, lauroyl peroxide, 2,2'-azobisisobutyronitrile, 2,2'-azobis-(2,4-dimethylvaleronitrile), o-chlorobenzoyl peroxide and o-methoxybenzoyl peroxide.

The dispersion and emulsion stabilizers usable in the present invention are those usually used. They include, for example, water-soluble polymers such as starch, hydroxyethylcellulose, carboxymethylcellulose, polyvinylpyrrolidone, polyvinyl alkyl ethers and polyvinyl alcohols; difficultly water-soluble inorganic salts such as barium sulfate, calcium sulfate, barium carbonate, calcium carbonate, magnesium carbonate and calcium phosphate; and surfactants such as sodium lauryl sulfate, sodium cetyl sulfate and sodium polyoxyethylene lauryl ether sulfate.

The monomer solution comprising the monomer(s), diluent (an agent for making the polymer porous) and polymerization initiator is dispersed or suspended in water in the presence of the above-mentioned dispersion or emulsion stabilizer by a known process to form droplets and then polymerization is conducted to prepare the intended porous polymer. In this process, the monomer solution is dispersed or suspended by an ordinary method such as an ordinary stirring method with any of various ordinary stirrers or a forced stirring method with a homogenizer or by irradiation with ultrasonic waves. These means are selected suitably to obtain particles having intended particle diameters. The proper polymerization temperature varies depending on the polymerization initiator used. It is usually in the range of 20 to 95°C. As a matter of course, the polymerization temperature must be lower than the boiling points of the monomer(s) and diluent under atmospheric pressure.

After the completion of the polymerisation, the polymer particles are separated by filtration. The aqueous phase is removed. The polymer particles are washed with water and/or with a solvent in order to replace the diluent having a high boiling point with the solvent having a lower boiling point. The product is then pulverised by an ordinary method such as spray-drying or reduced-pressure drying.

Before the polymerization, the mixture is in the form of droplets of a homogeneous solution comprising the monomer(s), diluent and initiator. As the polymerization proceeds, however, the diluent which is a poor solvent for the polymer(s) causes microscopic phase separation and then vaporized (desorbed) during drying under reduced pressure to form the intended porous polymer.

The oil absorption of the oil-absorptive polymer is determined by the "Method of Measurement of Oil Absorption of Pigment" stipulated in JIS K 5101 (1978) using squalene as the oil. In this method, 1 g of the powder is placed on a glass plate and kneaded by means of a spatula while squalene is gradually added dropwise thereto until the powder is wholly converted into a paste. The amount of squalene (ml) required per gram of the powder is referred to as the absorption. Although boiled linseed oil is used in JIS, squalene is used in the process of the present invention, since it is close to sebum. The absorption of the polymer according to the present invention is at least 1 ml/g, preferably at least 2 ml/g. The porosity can be determined in terms of the pore volume determined from the pore distribution measured by mercury porosimetry. The detail of this method is described in, for example, "Funtai Kogaku Binran" edited by Funtai Kogaku-kai and published by Nikkan Kogyo Shinbun-sha in 1986. The pore volume is at least 0.1 ml/g, preferably at least 0.15 ml/g.

The porous oil-absorptive polymer thus prepared can be dispersed in water, a lower alcohol or a mixture of them to form a dispersion usable as a hair lotion. The most preferred base of the hair preparation of the present invention is a mixture of water with ethanol in a weight ratio of 99/1 to 20/80, preferably 95/5 to 40/60. The amount of the oil-absorptive polymer in the hair lotion is preferably 0.05 to 5 wt.%, particularly

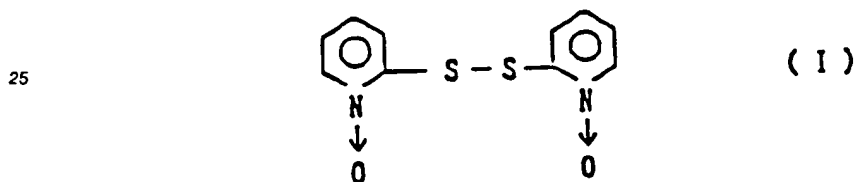
0.1 to 2 wt.%.

The lotion can contain further various water-soluble polymers in order to stabilize the dispersion or to dress the hair. They include monionic water-soluble polymers such as polyvinyl alcohol, polyvinylpyrrolidone and hydroxyethylcellulose; and anionic water-soluble polymers such as carboxymethylcellulose, crosslinked polyacrylic acid (carboxyvinyl polymer), xanthan gum and guar gum. These polymers are used in an amount of preferably 0.01 to 5 wt.%, particularly 0.05 to 1 wt.%.

The hair preparation of the present invention may contain, in addition to the above-described components, such other ordinary components in such amounts that they do not damage the effect of the present invention. They include, for example, polyhydric alcohols, e.g. glycerol, dipropylene glycol and 1,3-butylene glycol, nonionic surfactants, cationic surfactants, anionic surfactants, dyes, antiseptics, flavors, antioxidants, chelating agents, germicides, medicinal preparations, e.g. vitamins and hormones, astringents and U.V. absorbers.

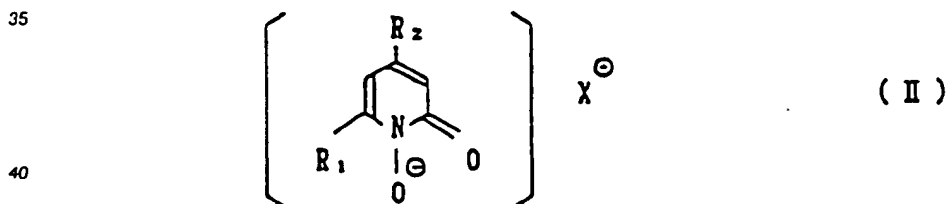
The porous vinyl polymer contained in the hair preparation of the present invention has a far higher capacity of absorbing squalene than that of ordinary inorganic powders and water-insoluble polymers. Therefore, the hair preparation containing the porous vinyl polymer exhibits an excellent effect of reducing the greasiness as will be shown in the following Examples.

Anti-dandruff agents which may be incorporated into the hair preparation of the present invention are well known ones. They include, for example, polyvalent metal salts of 2-mercaptopyridine N-oxide, colloidal sulfur, sulfur-containing amino acids and salts of them described in Japanese Patent Laid-Open No.183614/1983, 2,2'-dithiobispyridine 1,1'-dioxide of the formula (I):

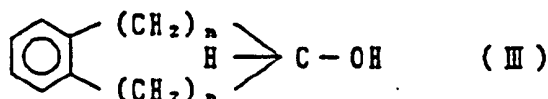


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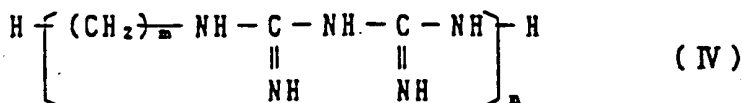
and hydrates of metal salts thereof such as magnesium sulfate thereof, 1-hydroxy-2-pyridone derivatives of the general formula (II):



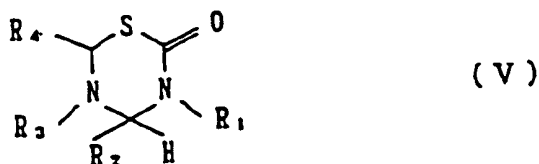
wherein R<sub>1</sub> represents an alkyl group having 1 to 17 carbon atoms, an alkenyl group having 2 to 17 carbon atoms, a cycloalkyl group having 5 to 8 carbon atoms, a bicycloalkyl group having 7 to 9 carbon atoms, a cycloalkylalkyl group in which the alkyl group has 1 to 4 carbon atoms and the cycloalkyl group may be substituted with an alkyl group having 1 to 4 carbon atoms, an aryl group, an aralkyl group in which the alkyl group has 1 to 4 carbon atoms, an arylalkenyl group in which the alkenyl group has 2 to 4 carbon atoms, an aryloxyalkyl group or an arylmercaptoalkyl group in which the alkyl group has 1 to 4 carbon atoms, a benzhydryl group, a phenylsulfonylalkyl group in which the alkyl group has 1 to 4 carbon atoms, a furyl or furylalkenyl group in which the alkyl group has 2 to 4 carbon atoms (the aryl or furyl group in each of the above-mentioned groups may be substituted with an alkyl or alkoxy group having 1 to 4 carbon atoms, a nitro group, a cyano group or a halogen atom), R<sub>2</sub> represents a hydrogen atom, an alkyl group having 1 to 4 carbon atoms, an alkenyl or alkynyl group having 2 to 4 carbon atoms, a halogen atom, a phenyl group or a benzyl group, and X represents a residue of an organic amine, described in Japanese Patent Publication No. 39805/1983, salicylic acid and its derivatives, triethyl citrate described in Japanese Patent Laid-Open No. 180417/1983, indole derivatives of the general formula (III):



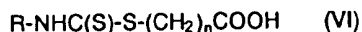
described in West German Patent No. 3142296, compounds of the general formula (IV):



wherein  $m$  represents a number of 1 to 10 and  
 $n$  represents a number of 1 to 6,  
 and salts of them described in West German Patent No. 012767, 2-oxotetrahydro-1,3,5-thiadiazine derivatives of the general formula (V):



described in west German Patent No. 3022799,  $\omega$ -(aminothiocabonylmercapto)alkanoic acids described in  
 Belgian Patent No. 2085728, compounds of the general formula (VI):



and salts of them, quinone derivatives, selenium disulfide, phenol derivatives and coal tar.

Among them, the polyvalent metal salts of 2-mercaptopyridine, 2,2'-dithiobispyridine 1,1'-dioxide of the formula (I) and 1-hydroxy-2-pyridone derivatives of the general formula (II) are preferred from the viewpoints of the safety and effectiveness. Particularly preferred are zinc salt of 2-mercaptopyridine (zinc pyrithione), triethanolamine salt of 1-hydroxy-4-methyl-6-(2,4,4-trimethylpentyl)-2(1H)-pyridone (Piroctone Auramine) and 2,2'-dithiobispyridine 1,1'-dioxide. The zinc salt of 2-mercaptopyridine is used preferably in the form of fine particles at least 50 wt.% of which have a particle diameter of 0.2  $\mu\text{m}$  or less and which have an average particle diameter of 0.2  $\mu\text{m}$  or less as described in Japanese Patent Laid-Open Nos. 16972/1985, 16973/1985 and 224676/1985.

The oil absorption of the oil-absorptive fine particles is determined by the "Method of Measurement of Oil Absorption of Pigment" according to JIS K 5101 (1978) using squalene as the oil in the present invention. In this method, 1 g of the powder is placed on a glass plate and kneaded by means of a spatula while squalene is gradually added dropwise thereto until the powder is wholly converted into a paste. The amount of squalene (mL) required per gram of the powder is referred to as the absorption. Although boiled linseed oil is used in JIS, squalene is used in the process of the present invention, since it is close to sebum.

The oil absorption of typical, commercially available oil-absorptive substances determined by the above-mentioned method is shown for comparison purposes in Table 1, wherein squalene is used as the oil.



Table 1

	Oil-absorptive substance	Oil absorption (ml/g)
5	synthetic phyllosilicate [ $\text{Mg}_3(\text{Si}_4\text{O}_{11})\text{H}_2\text{O} \cdot 11\text{H}_2\text{O}$ ]	1.07
	ditto [ $\text{K}_2\text{O} \cdot 3\text{Al}_2\text{O}_3 \cdot 6\text{SiO}_2$ ]	0.99
	synthetic platy silicate [ $\text{Ca}_4(\text{Si}_6\text{O}_{16})(\text{OH}) \cdot 4\text{H}_2\text{O}$ ]	4.96
	ditto [ $\text{Ca}_4(\text{Si}_6\text{O}_{16})(\text{OH})(\text{OH})$ ]	5.09
	dry synthetic platy silicate [ $\text{Ca}(\text{Si}_6\text{O}_{16})(\text{OH})$ ]	1.40
10	porous colloidal silica having a pore diameter of 120 Å [ $\text{SiO}_2 \cdot n\text{H}_2\text{O}$ ]	1.52
	porous colloidal silica having a pore diameter of 60 Å [ $\text{SiO}_2 \cdot n\text{H}_2\text{O}$ ]	1.38
	Talc JA46A	0.54
	Kaolin A	0.72
	bentonite mainly comprising montmorillonite	0.86
15	spherical colloidal silica [Aerosil 200]	5.00
	spherical colloidal silica [Aerosil 130]	5.38
	porous nylon beads [Orgasol 2000D]	0.80
	non-porous nylon beads [Toray Nylon SP500]	0.57

20 The hair preparation of the present invention may be in either solid or liquid form. Usually a liquid suspension of the oil-absorptive substance in the form of fine particles is preferably used. The liquid in which the anti-dandruff agent and the oil-absorptive fine particles of the present invention are suspended is, for example, water, a lower alcohol or a mixture of them. When the hair preparation in solid form is applied to the scalp and hair and rubbed thereinto, it becomes liquid to uniformly and effectively spread the anti-dandruff agent and the oil-absorptive agent on the surfaces thereof. The solid hair preparation is desirably in the form of gel, paste or the like. It can be used easily when it contains a substantial amount of a solvent.

25 The alcohols used as the solvent are preferably lower monohydric alcohols and part of them may be replaced with a dihydric or polyhydric alcohol. The monohydric alcohols usable in the present invention are preferably ethanol and isopropanol. Mixtures of two or more components selected from the group consisting of ethanol, isopropanol, water and other monohydric, dihydric and polyhydric alcohols can also be used. The anti-dandruff agent can be suspended or dissolved also therein.

30 To keep the suspension of the oil-absorptive substance homogeneous, an organic gum can be used. The organic gums include, for example, carboxymethylcellulose, hydroxyalkylcellulose, carboxypolymethylene, polyvinylpyrrolidone, acrylates, gelatin, dextrin, tragacanth, acacia, alginic acid, pectin and carrageenan.

35 The hair preparation of the present invention may contain various other components for various purposes. They include, for example, ordinary colorants such as dyes and pigments, flavors, vitamins, hair growth promoting agents and surfactants.

40 The hair preparation of the present invention is a new one prepared on the basis of a new conception unlike ordinary anti-dandruff compositions. The hair preparation is effective in preventing dandruff and itch and also in preventing hair from becoming greasy, which were problems of the ordinary anti-dandruff compositions.

#### 45 [Examples]

The following Preparation Examples and Examples will further illustrate the present invention, which by no means limit the invention.

#### 50 Preparation Example 1

40 g of styryl methacrylate, 40 g of divinylbenzene, 80 g of toluene, 2 g of benzoyl peroxide and 700 g of a 0.6% aqueous polyvinyl alcohol solution were placed in a 2-l separable flask and mixed by means of a homomixer at 11,000 rpm for 5 min. Then the dispersion thus obtained was stirred at 300 rpm at 80 °C for 8 h in a nitrogen atmosphere to conduct polymerization. After the completion of the polymerization, the reaction mixture was filtered. The filtration residue was washed with water and then with acetone and dried to obtain 40 g of a porous vinyl polymer having an average particle diameter of 6.7 μm.

55 According to the results of the determination of the pore distribution by mercury porosimetry, the polymer had pores of smaller than 347 Å on their particle surfaces and a pore volume of 0.15 ml/g.

## Preparation Example 2

20 g of stearyl methacrylate, 20 g of styrene, 40 g of divinylbenzene, 30 g of toluene, 90 g of n-dodecane, 0.8 g of benzoyl peroxide and 470 g of a 3.3% aqueous sodium dodecylsulfate solution were placed in a 2-l separable flask and stirred at 300 rpm at 65°C for 9 h in a nitrogen atmosphere to conduct polymerization. After the completion of the polymerization followed by the same procedure as that of Preparation Example 1, 50 g of a porous vinyl polymer having an average particle diameter of 2.0  $\mu\text{m}$  was obtained.

According to the results of the determination of the pore distribution by mercury porosimetry; the polymer had pores of smaller than 725 Å on their particle surfaces and a pore volume of 0.299 ml/g.

## Preparation Example 3

17 g of stearyl methacrylate, 17 g of divinylbenzene, 34 g of toluene, 1 g of potassium persulfate and 530 g of 1.0% aqueous polyoxyethylene (35 mol of ethylene oxide added) nonylphenyl ether were placed in a 2-l separable flask and stirred at 300 rpm at 75°C for 8 h in a nitrogen atmosphere to conduct polymerization. After the completion of the polymerization followed by topping conducted under 100 to 150 mmHg for 2 to 3 h, a porous vinyl polymer was obtained in the form of an aqueous suspension. The average particle diameter of the polymer was smaller than 0.7  $\mu\text{m}$ .

## Preparation Example 4 (Comparative)

A porous vinyl polymer having an average particle diameter of 2.5  $\mu\text{m}$  was prepared in the same manner as that of Preparation Example 2 except that the 20 g of styrene was replaced with 20 g of t-butylstyrene and the amount of divinylbenzene was altered from 40 g to 0.4 g.

## Preparation Examples 5 and 6

A vinyl polymer was prepared in the same manner as that of Preparation Example 1 except that the amount of toluene was altered to 160 g (Preparation Example 5) or no toluene was used (Preparation Example 6).

The polymer thus prepared in Preparation Example 6 was a comparative one.

The oil absorptivity of the polymers prepared in the above Preparation Examples 1 to 6 was determined according to the above-mentioned method based on JIS K 5101. For comparison, the oil absorptivity of talc, kaolin and Polymer No. 14 in Example 1 of U.S. Patent No. 4489058 which has the highest oil absorptivity among all the polymers mentioned in the specification of this patent and which comprises 65 wt.% of t-butylstyrene, 35 wt.% of stearyl methacrylate and 0.0125 wt.% of divinylbenzene as the crosslinking agent was determined.

The results are shown in Table 2.

Table 2

	Squalene absorption (ml/g)
Preparation Example 1	2.0 $\pm$ 0.1
Preparation Example 2	3.3 $\pm$ 0.3
Preparation Example 3	3.0 $\pm$ 0.1
Preparation Example 4	2.5 $\pm$ 0.1
Preparation Example 5	2.9 $\pm$ 0.1
Preparation Example 6	0.6 $\pm$ 0.1
polymer No. 14 of U.S. patent 4 489 058	0.6 $\pm$ 0.1
talc	0.5 $\pm$ 0.1
kaolin	0.5 $\pm$ 0.1

## Comparative Example 1

Comparative Scalp Lotions A, B, C and D (controls) having compositions shown in Table 3 were prepared.

- 5 Piroctone Auramine [1-hydroxy-4-methyl-6-(2,4,4-trimethylpentyl)-2(1H)-pyridone triethanolamine salt] is an anti-dandruff agent of Hoechst (trade name: Octopirox). It has been commercially used in the fields of shampoos, rinses and hair tonics and its effect has already been recognized. Bentonite is an inorganic bedded clay mineral mainly comprising montmorillonite and usable as a carrier having a relatively high oil absorptivity. Bentonite having a primary particle diameter of around 2  $\mu\text{m}$  (sebum absorption: 0.86 ml squalene/g) was used. The spherical primary particles of the colloidal silica used had a particle diameter of  
 10 around 12 nm (sebum absorption: 5 ml squalene/g).

Table 3

15	Formulation	Scalp lotion A (present invention)	Scalp lotion B (present invention)	Scalp lotion C (comp.)	Scalp lotion D (comp.)
20	bentonite mainly comprising montmorillonite	0.2 wt%	-	-	-
	colloidal silica (Aerosil 200)	-	0.2	-	-
	Piroctone Auramine	0.2	0.2	0.2	-
25	lactic acid	proper amt.	proper amt.	proper amt.	proper amt.
	flavor	proper amt.	proper amt.	proper amt.	proper amt.
	hydroxyethylcellulose	0.05	-	0.05	0.05
	ethanol	50	-	50	50
30	isopropanol	-	50	-	-
	purified water	balance	balance	balance	balance
	pH adjustment	6 ~ 7	6 ~ 7	6 ~ 7	6 ~ 7

- 35 Ten male subjects (aged 18 to 25) previously found to be suffering from dandruff and itch were subjected to clinical tests using the scalp lotions A, B, C and D for 6 weeks.

- In the clinical tests, the scalp lotion D (control) was used in the first and second weeks, the scalp lotion C was used in the third and fourth weeks and the scalp lotion A (for five subjects in group 1) or scalp lotion B (for five subjects in group 2) was used in the fifth and sixth weeks. In practice, they shampooed with the control shampoo three times a week (on Monday, Wednesday and Friday). In each case, after shampooing with a control shampoo followed by drying with a towel, about 4 g of the scalp lotion was spread over the scalp. On Monday after two weeks, the formation of the dandruff and the development of greasiness of the hair were judged by the panellists and classified into four ranks. The results are shown in Table 4.

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Table 4

Scalp lotion	Dandruff	Grasiness
D	2.8	1.8
C	0.6	2.8
B	0.4	0.8
A	0.4	0
Notes) 0: scarcely observed 1: slightly observed 2: observed 3: considerably observed.		

## Example 2 (Comparative):

Scalp treatments (to be used in a shampoo-free condition) A and B of the present invention and comparative scalp treatments C, D and E having the compositions shown in Table 5 were prepared.

Table 5

Formulation					
	A	B	C	D	E
porous vinyl polymer prepared in Preparation Example 4	0.3 wt%	0.3	0.3	-	-
Carbopol 941 <sup>*1</sup>	0.18	0.18	0.18	0.18	0.18
Silicone KF352A <sup>*2</sup>	0.1	0.1	0.1	0.1	0.1
triethanolamine	proper amt.	proper amt.	proper amt.	proper amt.	proper amt.
2-amino-2-methyl-1-propanol	proper amt.	proper amt.	proper amt.	proper amt.	proper amt.
Octopirox <sup>*3</sup>	0.1	-	-	-	-
zinc pyrithione (Zpt)	-	0.1	-	0.1	-
ethanol	50	50	50	50	50
flavor	proper amt.	proper amt.	proper amt.	proper amt.	proper amt.
purified water	balance	balance	balance	balance	balance
pH control	6 ~ 7	6 ~ 7	6 ~ 7	6 ~ 7	6 ~ 7

Notes)

<sup>\*1</sup>: carboxyvinyl polymer (a product of Goodrich)<sup>\*2</sup>: polyether-modified silicone<sup>\*3</sup>: Piroctone Auramine (a product of Hoechst)

The comparative scalp treatments A to E were subjected to the same clinical test as that of Example 1 to examine the effects of them in controlling the formation of dandruff and greasiness of the hair. The results are shown in Table 6.

Table 6

Scalp treatment	Dandruff	Greasiness
A	0.8	0.2
B	0.2	0.2
C	2.8	1.0
D	0.2	3.0
E	3.0	2.2

## Example 3

Comparative hair conditions A and B having the compositions shown in Table 7 were prepared.

Table 7

Formulation	A	B
porous vinyl polymer prepared in Preparation Example 4	4.5 wt%	-
Octopirox	0.3	0.3
dialkyldimethylammonium chloride	0.5	0.5
stearyltrimethylammonium chloride	1.0	1.0
solid paraffin	0.5	0.5
liquid paraffin	0.5	0.5
polymethylsiloxane	0.5	0.5
methycellulose	0.5	0.5
propylene glycol	3.0	3.0
stearyl alcohol	3.0	3.0
colorant	proper amt.	proper amt.
flavor	proper amt.	proper amt.
purified water	balance	balance

## Comparative Example 4

Comparative shampoos A and B having the compositions shown in Table 8 were prepared.

Table 8

Formulation	A	B
fine particles of zeolite <sup>*1</sup>	5.0 wt%	-
Zpt	1.0	1.0
AS-TEA <sup>*2</sup>	20	20
Carbopol 941	0.5	0.5
urea	5.0	5.0
triethanolamine	2.0	2.0
thickening agent (Cellosize QP 52000H)	0.1	0.1
pigment	proper amt.	proper amt.
flavor	proper amt.	proper amt.
purified water	balance	balance

Notes)

\*1: average particle diameter: 0.8  $\mu\text{m}$  (prepared by pulverizing ordinary zeolite with a sand grinder)

\*2: triethanolamine laurylsulfate

## Example 5

Hair preparations A, B and C of the present invention having the compositions shown in Table 9 were prepared.

Table 9

	Present invention		
	A	B	C
porous vinyl polymer prepared in Preparation Example 1	0.3 wt%	-	-
porous vinyl polymer prepared in Preparation Example 2	-	0.3	-
porous vinyl polymer prepared in Preparation Example 3	-	-	0.3
Carbopol 941	0.04	0.04	0.04
1-menthol	0.05	0.05	0.05
2,2'-dithiobispyridine	0.1	0.1	0.1
1,1'-dioxide	0.1	0.1	0.1
flavor	0.2	0.2	0.2
thanol	48	48	48
water	balance	balance	balance

In Examples 3 to 5, the preparations of the present invention containing both the anti-dandruff agent and the oil-absorptive substance exhibited both the anti-dandruff and greasiness reducing effects.

## Example 6

A pasty hair preparation having the following composition was prepared:

5	porous vinyl polymer prepared in Preparation Example 1	15 wt.%
	carboxymethylcellulose	0.5
	ethyl alcohol	30
	flavor	0.1
10	water	q.s. ad 100 wt.%

For comparison, pasty hair preparations were prepared in the same manner as above except that the polymer prepared in Preparation Example 1 was replaced with Polymer No. 14 described in the specification of U.S. Patent No. 4,489,058, talc or kaolin.

15 Each preparation was applied to the cheek of a subject and peeled off after one hour. The amount of sebum thus absorbed was determined by gravimetry. The results are shown in Table 10.

Table 10

20		Preparation of the present invention	Polymer No. 14 of U.S. Patent No. 4,489,058	Preparation comprising talc	Preparation comprising kaolin
25	Amount of sebum ( $\mu\text{g}/\text{cm}^2$ )	$18 \pm 6$	$2 \pm 1$	$12 \pm 8$	$8 \pm 6$

It is apparent from Table 10 that the preparation of the present invention has a higher capacity of absorbing sebum than the comparative ones.

## Example 7

A hair preparation having the following composition was prepared:

35	water-insoluble, porous vinyl polymer prepared in Preparation Example 2	0.3 wt.%
	carboxyvinyl polymer (Carbopol 941; a product of Goodrich)	0.04
	l-menthol	0.05
	flavor	0.2
	ethyl alcohol	48
40	purified water	q.s. ad 100 wt.%

The hair preparation was applied to the hair on one side of the head of each of six subjects and a polymer-free preparation (comparative preparation) was applied to the hair on the other side. A difference in the greasiness between the both sides was examined organoleptically by the monitors after one, two and three days.

45 The results are shown in Table 11.

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Tabl 11

Subject	After 1 day		After 2 days		After 3 days	
	Present invention	Comp.	Present invention	Comp.	Present invention	Comp.
A	○	△	○	x	○	x
B	○	○	○	○	○	△
C	○	○	○	△	△	x
D	○	△	○	x	△	x
E	○	○	△	x	x	x
F	○	○	○	△	△	△
Notes) ○: nearly free from greasiness △: slight greasiness x: strong greasiness						

## Example 8

A hair conditioner having the following composition was prepared:

dispersion of porous vinyl polymer prepared in Preparation Example 3 (in terms of solid)	1.0 wt. %
stearyltrimethylammonium chloride	1.5
cetanol	2.3
propylene glycol	5.0
methylparaben	0.2
hydroxyethylcellulose	0.8
Blue No. 1	a suitable amount
flavor	ditto
purified water	the balance

The hair conditioner was applied to the hair on one side of the head of each of five subjects and a polymer-free comparative preparation was applied to the hair on the other side. After washing with water, the greasiness was examined after one, two and three days in the same manner as that of Example 7.

The results are shown in Table 12.

Table 12

Subject	After 1 day		After 2 days		After 3 days	
	Present invention	Comp.	Present invention	Comp.	Present invention	Comp.
A	○	○	○	x	○	x
B	○	○	○	△	△	x
C	○	○	○	△	○	△
D	○	○	△	△	△	x
E	○	△	x	x	x	x

## Claims

1. A hair preparation characterized in that it comprises particles of an oil-absorptive polymer selected from polymethacrylic stearate, polyacrylic stearate, a methacrylic stearate/styrene copolymer and an acrylic stearate/styrene copolymer having a solubility parameter of 7 to 10, said particles having an average



particle diameter of 0.0005 to 30  $\mu\text{m}$  (microns) and a pore volume of at least 0.1 ml/g, said polymer having a squalene absorption of at least 1 ml/g, and a carrier.

2. A hair preparation as claimed in Claim 1, which comprises 0.05 to 5 percent by weight of the oil-absorptive polymer and the balance of the carrier.
3. A hair preparation as claimed in Claim 1, in which said oil-absorptive polymer has an average particle size of 0.0005 to 30  $\mu\text{m}$  (microns) and has such a size distribution that it contains 90 percent by weight or more of those having particle sizes in the range of 0.0005 to 30  $\mu\text{m}$  (microns).
4. A hair preparation as claimed in any preceding claim, wherein the polymer is a copolymer of methacrylic stearate and styrene the weight ratio of methacrylic stearate : styrene being not less than 30 : 70.
5. A hair preparation according to any of claims 1 to 3 wherein the polymer is a copolymer of methacrylic stearate and styrene having a glass transition point ( $T_g$ ) of 100 °C or higher.
6. A hair preparation according to any of claims 1 to 3 wherein the polymer is a copolymer of methacrylic acid and styrene substituted with a  $C_1$  to  $C_{12}$  alkyl group.

#### Patentansprüche

1. Haarpräparat, dadurch gekennzeichnet, daß es Teilchen eines ölabSORbierenden Polymers, ausgewählt aus Polymethacrylstearat, Polyacrylstearat, einem Methacrylstearat/Styrol-Copolymer und einem Acrylstearat/Styrol-Copolymer mit einem Löslichkeitsparameter von 7 bis 10, wobei die Teilchen einen durchschnittlichen Teilchendurchmesser von 0,0005 bis 30  $\mu\text{m}$  (microns) und ein Porenvolumen von wenigstens 0,1 ml/g besitzen, und das Polymer eine Squalenabsorption von wenigstens 1 ml/g besitzt, und einen Träger umfaßt.
2. Haarpräparat nach Anspruch 1, welches 0,05 bis 5 Gew.-% des ölabSORbierenden Polymers und als Rest den Träger umfaßt.
3. Haarpräparat nach Anspruch 1, wobei das ölabSORbierende Polymer eine durchschnittliche Partikelgröße von 0,0005 bis 30  $\mu\text{m}$  (microns) besitzt, und solch eine Größenverteilung aufweist, daß es 90 Gew.-% oder mehr an solchen mit Teilchengrößen im Bereich von 0,0005 bis 30  $\mu\text{m}$  (microns) enthält.
4. Haarpräparat nach einem der vorhergehenden Ansprüche, wobei das Polymer ein Copolymer von Methacrylstearat und Styrol ist und das Gewichtsverhältnis von Methacrylstearat : Styrol nicht weniger als 30 : 70 beträgt.
5. Haarpräparat nach einem der Ansprüche 1 bis 3, wobei das Polymer ein Copolymer von Methacrylstearat und Styrol mit einem Glasübergangspunkt ( $T_g$ ) von 100 °C oder höher ist.
6. Haarpräparat nach einem der Ansprüche 1 bis 3, wobei das Polymer ein Copolymer von Methacrylsäure und Styrol, substituiert mit einer  $C_1$ - bis  $C_{12}$ -Alkylgruppe, ist.

#### Revendications

1. Préparation pour les cheveux, caractérisée en ce qu'elle comprend des particules d'un polymère absorbant l'huile choisi parmi un polyméthacrylate de stéaryle, un polyacrylate de stéaryle, un copolymère méthacrylate de stéaryle/styrène et un copolymère acrylate de stéaryle/styrène ayant un paramètre de solubilité de 7 à 10, lesdites particules ayant un diamètre moyen de particules de 0,0005 à 30  $\mu\text{m}$  (microns) et un volume poreux d'au moins 0,1 ml/g, ledit polymère ayant une capacité d'absorption de squalène d'au moins 1 ml/g, et un support.
2. Préparation pour les cheveux telle que revendiquée dans la revendication 1, qui comprend 0,05 à 5 pour cent en poids du polymère absorbant l'huile, et le complément du support.

3. Préparation pour les cheveux telle que revendiquée dans la revendication 1, dans laquelle ledit polymère absorbant l'huile a une taille moyenne de particules de 0,0005 à 30  $\mu\text{m}$  (microns) et une distribution granulométrique telle qu'il contienne 90 pour cent en poids plus de particules ayant des tailles comprises dans l'intervalle de 0,0005 à 30  $\mu\text{m}$  (microns).
- 5 4. Préparation pour les cheveux telle que revendiquée dans l'une quelconque des revendications précédentes, dans laquelle le polymère est un copolymère de méthacrylate de stéaryle et de styrène, le rapport en poids méthacrylate de stéaryle:styrène étant d'au moins 30:70.
- 10 5. Préparation pour les cheveux selon l'une quelconque des revendications 1 à 3, dans laquelle le polymère est un copolymère de méthacrylate de stéaryle et de styrène ayant une température de transition vitreuse ( $T_v$ ) de 100 °C ou plus.
- 15 6. Préparation pour les cheveux selon l'une quelconque des revendications 1 à 3, dans laquelle le polymère est un copolymère d'acide méthacrylique et de styrène substitué par un groupe alkyle en C<sub>1</sub>-C<sub>12</sub>.

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